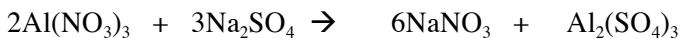
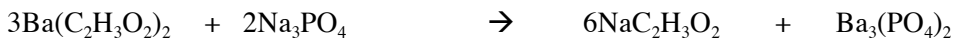


Some Review Equations!!

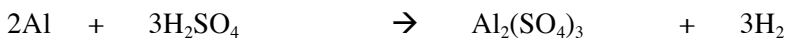
1. Aluminum nitrate<sub>(aq)</sub> + sodium sulfate<sub>(aq)</sub>



2. Barium acetate<sub>(aq)</sub> combines with sodium phosphate<sub>(aq)</sub>



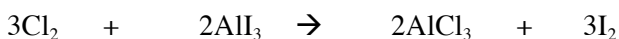
3. Aluminum is reacted with sulfuric acid<sub>(aq)</sub>



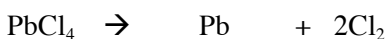
4. Magnesium is mixed with sodium chloride<sub>(aq)</sub>

No reaction. Mg will not replace sodium

5. Chlorine is mixed with aluminum iodide<sub>(aq)</sub>



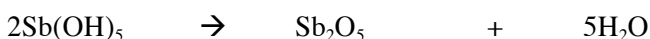
6. Lead (IV) chloride decomposes



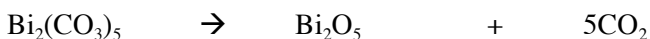
7. Lead (IV) chlorate decomposes



8. Antimony (V) hydroxide decomposes



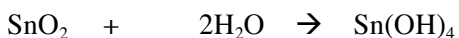
9. Bismuth (V) carbonate decomposes



10. Hydrochloric acid decomposes



11. Tin(IV) hydroxide is synthesized from a metal oxide and water



12. Aluminum metal is placed in a solution of copper (II) nitrate.

