The Three Formulas: Used to calculate energy change during:

1. $q = mC_p\Delta T$ temperature change

2. $q = mH_f$ melting or freezing

3. $q = mH_v$ boiling or condensing

The variables stand for:

q = quantity of heat m = mass C_p = Specific Heat Capacity Cp of Water = 4.18 J/gC

H_f = heat of fusion H_v = heat of vaporization ΔT = change in temperature

Cp of Water = 4.18 J/gC Hf of water = 334 J/g Hv of water = 2260 J/g

Problems:

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