

Formula Alt.

Write formulas for the following. Write symbols for each ion necessary to create the formula, as shown in the first problem.

Sodium oxide       $\text{Na}^{+1}$        $\text{Na}^{+1}$        $\text{O}^{-2}$        $\text{Na}_2\text{O}$

Potassium phosphide \_\_\_\_\_

Magnesium iodide \_\_\_\_\_

Barium oxide \_\_\_\_\_

Strontium chloride \_\_\_\_\_

Calcium chlorate \_\_\_\_\_

Lithium nitride \_\_\_\_\_

Aluminum oxide \_\_\_\_\_

Aluminum nitride \_\_\_\_\_

Potassium arsenide \_\_\_\_\_

Magnesium phosphate \_\_\_\_\_