

## Gay-Lussac's Law

Purpose: To gather experimental evidence of Gay-Lussac's law.

Materials: 125 ml Erlenmeyer flask                      Burner and Ring Stand  
Pressure and temperature sensors, interface, etc.    Large beakers

Procedure: Set up the system as demonstrated in class.

Open the Pressure-Temperature experiment from the Chemistry with Vernier folder.

Create a water bath at a temperature of your choice. **DO NOT USE A TEMPERATURE OVER 60 C.** *After the flask sits in the water bath for a period of 5 minutes*, use Logger Pro to record the temperature and pressure.

Create a graph of the temperature of the water and the pressure of the gas at each of at least **4 different temperatures (5 is better)**. Put the dependent variable on the correct axis.

- Data:
- Do a linear regression (a linear fit on Logger Pro) on the data. Do not force a zero.
  - Save your data and graph. **THIS IS GRAPH ONE.**
  - Click on the graph and labels and change them so that you have a graph with temperature on the Y axis and pressure on the X. This is **GRAPH TWO**. Again, do a linear regression.

### Conclusion

1. What does graph one tell you about the relationship between temperature and pressure?

2. On **GRAPH TWO**, what is the value of the Y intercept? \_\_\_\_\_

3. Using **GRAPH TWO**, as the pressure decreased, what was happening to the temperature? \_\_\_\_\_

4. Using **Graph TWO** what would be the value of the pressure when the line crossed the y axis? \_\_\_\_\_

5. Using **Graph TWO**, what is the value of T when the line crosses the y axis? \_\_\_\_\_

6. When the line crosses the y axis, there is 0 pressure. What causes pressure?

7. Since there is no pressure, is there temperature? (temperature is caused by movement of molecules) \_\_\_\_\_

8. What is the significance of the Y intercept on graph two? \_\_\_\_\_

9. What is your value for absolute zero? \_\_\_\_\_

10. What the value of absolute zero in degrees K? \_\_\_\_\_