

Some Equations!! Write and BALANCE

1. Aluminum nitrate_(aq) + sodium sulfate_(aq)

2. Barium acetate_(aq) combines with sodium phosphate_(aq)

3. Aluminum is reacted with sulfuric acid_(aq)

4. Magnesium is mixed with sodium chloride_(aq)

5. Chlorine is mixed with aluminum iodide_(aq)

6. Lead (IV) chloride decomposes

7. Plumbic chlorate decomposes

8. Antimony (V) hydroxide decomposes

9. Bismuth (V) carbonate decomposes

10. Hydrochloric acid decomposes

11. Chloric acid decomposes

12. Hypoiodous acid is synthesized from a nonmetal oxide and water.

13. Stannic hydroxide is synthesized from a basic anhydride (metal oxide) and water

14. Lead (IV) phosphate is synthesized from a basic anhydride (metal oxide) and an acid anhydride (nonmetal oxide).

15. Sodium oxide and dinitrogen trioxide are reacted.

16. Antimony (III) oxide and diarsenic pentoxide are reacted.