

## Formula Writing /Formula Mass Practice

Write Formulas and find the formula mass for the following compounds

FORMULA      MASS

1. hydrogen oxide      \_\_\_\_\_      \_\_\_\_\_

2. lithium sulfide      \_\_\_\_\_      \_\_\_\_\_

3. Aluminum chloride      \_\_\_\_\_      \_\_\_\_\_

4. sodium oxide      \_\_\_\_\_      \_\_\_\_\_

5. Barium chloride      \_\_\_\_\_      \_\_\_\_\_

6. Calcium sulfide      \_\_\_\_\_      \_\_\_\_\_

7. Aluminum phosphide      \_\_\_\_\_      \_\_\_\_\_

8. potassium nitride      \_\_\_\_\_      \_\_\_\_\_

9. potassium sulfide      \_\_\_\_\_      \_\_\_\_\_

10. potassium chloride      \_\_\_\_\_      \_\_\_\_\_

11. iron (II) bromide      \_\_\_\_\_      \_\_\_\_\_

12. calcium bromide      \_\_\_\_\_      \_\_\_\_\_

13. lithium phosphide      \_\_\_\_\_      \_\_\_\_\_

14. rubidium oxide      \_\_\_\_\_      \_\_\_\_\_

15. strontium fluoride      \_\_\_\_\_      \_\_\_\_\_

16. cobalt (III) sulfate      \_\_\_\_\_      \_\_\_\_\_

17. nickel (II) oxalate      \_\_\_\_\_      \_\_\_\_\_

18. lead(IV) chlorate      \_\_\_\_\_      \_\_\_\_\_

19. tin(II) chlorate      \_\_\_\_\_      \_\_\_\_\_

20. iron(III) iodate      \_\_\_\_\_      \_\_\_\_\_

21. aluminum oxalate      \_\_\_\_\_      \_\_\_\_\_

22. magnesium nitrate      \_\_\_\_\_      \_\_\_\_\_

23. bismuth(V) chlorate      \_\_\_\_\_      \_\_\_\_\_

24. aluminum sulfite      \_\_\_\_\_      \_\_\_\_\_

25. gallium chlorite      \_\_\_\_\_      \_\_\_\_\_

26. copper(I) carbonate      \_\_\_\_\_      \_\_\_\_\_

27. copper(II) sulfite      \_\_\_\_\_      \_\_\_\_\_

28. cesium oxide      \_\_\_\_\_      \_\_\_\_\_

29. nickel(II) sulfide      \_\_\_\_\_      \_\_\_\_\_

30. iron(II) phosphide      \_\_\_\_\_      \_\_\_\_\_